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Chapter 22: Hydrocarbon Compounds. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. Mary\_Brooks2. Terms in this set (36) hydrocarbon. contain only carbon and hydrogen, simplest organic compound. Because carbon has four valence electrons, a carbon atoms always forms \_\_\_\_ covalent bonds.

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CHAPTER 22, Hydrocarbon Compounds (continued) SECTION 22.4 HYDROCARBON RINGS (pages 709-711) This section describes how to identify and classify cyclic hydrocarbons. It also explains the bonding in benzene. Cyclic Hydrocarbons (page 709) 1. What is a cyclic hydrocarbon? It is a compound that contains a hydrocarbon ring. 2.

### This section describes the bonding in hydrocarbons and ...

GC Separation of Hydrocarbons A. Saturated and Unsaturated Aliphatic Hydrocarbon Compounds 1. C 1 -C 5 hydrocarbon compounds a. 50 m A1 2 O 3 /KCl Plot column. 40 (1 min)–200° at 10°/min or 60–200° at 3°/min. b. 30 to 50 m GS-Q column or Poraplot Q column. 35–100° at 10°/min. 100° (5 min); 100–200° at 10°/min, or 40 (2 min ...

### Hydrocarbon - an overview | ScienceDirect Topics

Some of these are listed in Table 22-7, which includes nitrogen, oxygen, and sulfur compounds, as well as hydrocarbons. Although petroleum from some locations contains fairly substantial amounts of aromatic hydrocarbons, it is not a principal source for such compounds.

### 22.11: Sources and Uses of Aromatic Hydrocarbons ...

hydrocarbons: organic compounds that contain only carbon and hydrogen: alkanes: hydrocarbons that contain only single covalent bonds: straight-chain alkanes: alkanes that contain any number of carbons one after another in a chain: substituent: atom or group of atoms that take the place of a hydrogen atom in a hydrocarbon molecule: alkyl group

### Quia - Chapter 22 "Hydrocarbon Compounds"

dioxide. Sulphur and oxygen forms sulphur di oxide. Hydrocarbons are compounds containing carbon and hydrogen. Methane is an example of hydrocarbon having the formula CH<sub>4</sub>. Ethane is a hydrocarbon and has a formula C<sub>2</sub>H<sub>6</sub>. Hydrocarbons are of straight chain compounds and branched chain compounds. We are dealing with only straight chain compounds.

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The number of carbon atoms as well as the bond type determines the name of the hydrocarbon. Explain how to name the organic compound based on the number of carbons and type of bond. The students will be reminded to look back at the chart provided to them regarding the prefixes and suffixes of hydrocarbons in order to identify hydrocarbons.

**Lesson Plans: Hydrocarbons (Senior, Science)**

An alkene is a hydrocarbon containing a double carbon-carbon bond., Alkyl Group: Any of a series of univalent groups of the general formula  $C_nH_{2n+1}$  derived from aliphatic hydrocarbons, Alkynes: Are hydrocarbons that have a triple bond between two carbon atoms, with the formula  $C_nH_{2n-2}$ , Aromatic Compound

**Quia - Chap 22 - Hydrocarbon Compounds**

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**Chapter 22 - Hydrocarbon Compounds - 22.5 Hydrocarbons ...**

Hydrocarbon derivatives are compounds that are made primarily of carbon and hydrogen atoms with specific groups of atoms attached. These specific groups of atoms are called functional groups .

**Hydrocarbon Derivatives: Definition & Properties - Video ...**

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