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Section 7.1 Ion Formation pages  
206–209 Section 7.1 Assessment page  
209 1. Compare the stability of a lithium  
atom with that of its ion,  $\text{Li}^+$ . The  $\text{Li}^+$  ion  
is more stable because it has a complete  
octet. 2.

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Answer : Octane number is the % age by volume of iso-octane and n-heptane which has the same knocking property as the fuel being tested under the similar experimental conditions in the special test engine. If we have greater % age of iso-octane the octane number is improved.

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